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# Modeling of diffusion filtration combustion radiative burner

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#### Abstract

Non-premixed filtration combustion is a new branch of the filtration combustion that combines some properties of turbulent diffusion flames and premixed filtration combustion. Two-dimensional, two-temperature and multi-component model for non-premixed filtration combustion simulation is presented. Parametric study of radiative efficiency of the burner is performed. The lower limit of total equivalence ratio  $\varphi$  for the simulated system is studied parametrically. Simulation predicts high stability of the flame at low equivalence ratios. It is shown that Burke–Schumann combustion and mixing rate well conforms to heat release rate in the non-premixed filtration combustion system. © 2001 Elsevier Science Ltd. All rights reserved.

# 1. Introduction

Gas combustion in a porous media or filtration combustion (FC) is under intensive investigations now due to a large number of technical applications such as VOCs oxidation, burning lean combustible mixtures under superadiabatic conditions, catalyst treatment by thermal wave, etc. [1,2]. Recently problems of flame front instability, transition between different regimes of the FC and others were addressed [3,4]. In spite of the progress in investigation of FC of premixed fuels the problem of non-premixed FC was not highlighted in the literature. We suppose to use the term filtration combustion in the case when Darcy or Forchheimer filtration hydrodynamics takes place and flame structure is affected by filtration hydrodynamics and interphase heat exchange. Usually this corresponds to small pore size, compared to the flame width (of the order of millimeters). In the opposite case the terms "sinked" flame or "bulked chamber" combustion is more appropriate.

Non-premixed filtration combustion (NFC) is a new area in FC. By analogy with diffusion flames it can be also termed as diffusion filtration combustion. This type

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of combustion combines some properties of diffusion turbulent flames (high stability and intensity of combustion, flame length conservation, etc.) and of premixed filtration combustion – low  $NO_x$ , CO emission level, internal heat recuperation, gas–solid temperature split and others.

One of the possible applications of NFC is radiative burner with controllable heat release and flame length. Since heat release rate is controlled by fuel and oxidant mixing in high temperature systems, the length of the flame L practically does not depend on the gas flow rate. Actually, the length of diffusion mixing is defined by system diameter  $d_2$ , gas velocity  $u_{\rm g}$  and diffusivity D. The effective diffusivity has dispersion nature and is proportional to the filtration velocity and porous media particle size  $D \sim u_{\rm g} d_0$  [5] and consequently

$$L \sim rac{d_2^2 u_{
m g}}{D} \sim rac{d_2^2 u_{
m g}}{d_0 u_{
m g}} = rac{d_2^2}{d_0}.$$

In this paper [5] the problem of coaxial gas flux mixing in condition of filtration was considered as a part of a NFC problem. Mixing rate was shown not to depend on total flow rate and variation of the granule size may be utilized for control of heat release rate and flame length. Although heat release rate is in good conformity with simple mixing model, development of the applications such as radiative burner requires more detailed heat and mass transfer simulation with realistic boundary conditions.

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Nomenclature		$T_{ m ext}$ $\mathbf{u}_{f g}$	ambient temperature, K filtration velocity vector, m/s
$egin{array}{ccc} C_i & & & & & & & \\ C_i & & & & & & & \\ C_i & & & & & & \\ C_i & & & & & & \\ D_d & & & & & \\ D_d & & & & & \\ d_0 & & & & & \\ d_1 & & & & & \\ d_2 & & & & & \\ G & & & & & \\ g_1, g_2 & & & & \\ H & & & & & \\ K, \widetilde{k} & & & & \\ L & & & & & \\ M & & & & & \\ m & & & & & \\ m & & & & & \\ p_r & & & & & \\ p_o & & & & \\ R & & & & & \\ \end{array}$	molar concentration of the <i>i</i> th gas component mass concentration gas diffusivity tensor dispersion diffusivity tensor longitudinal component of dispersion diffusivity transverse component of dispersion diffusivity gas diffusivity coefficient, cm²/s bedding particle diameter, m diameter of the internal tube, m gas flow rate, m³/h parameters in Eqs. (14) and (15) methane heat content, J/kg filtration permeabilities burner length, m gas molecular weight, kg/mol porosity normal vector to reactor boundary Prandtl number pressure, Pa outlet pressure, Pa gas constant		heat exchange coefficient,  W/K/m³ emissivity of the imbedded particles emissivity of external tube surface gas energy conductivity,  W/m/K gas viscosity, Pa s adiabate constant of the gas equivalence ratio heat conductivity tensor density, kg/m³ mass generation of ith component mass generation average over cross-section, kg/s/m  Stefan—Boltzmann constant radiative efficiency normalized velocity vector axial and radial components of the velocity vector projection
r	system radius, m	Subsc	ripts
$r_1$	radius of the internal tube, m	1	relates to internal tube or fuel
$r_2$	radius of the external tube or radius of the	2	relates to external tube or oxidant
	system, m	g i	gas
T	temperature, K	i	ith component of a gas
$T_0$	initial temperature of the system, K	S	solid

The aim of this paper is to simulate non-premixed filtration combustion radiative burner, explore the thermal characteristics of the tubular burner with coaxial feed and to demonstrate the possibility of heat release control via two-dimensional simulation.

# 2. Model of non-premixed filtration combustion burner with coaxial feed

Consider the following model of the non-premixed filtration combustion burner. A gas mixture consisting of different components is pumped through a cylindrical chamber of radius  $r_2$  and length L, filled with the inert fill of porosity m. Gas flow through the chamber is specified by flow rate G at the inlet cross-section (z = 0) and pressure  $p_0$  at the outlet cross-section (z = L) of the chamber. Inlet cross-section is divided into two parts ( $r < r_1$ ) and ( $r_1 < r < r_2$ ) through which gases of different chemical composition (usually fuel and oxidant) and velocity are fed.

$$G = \begin{cases} G_1, & r < r_1, \\ G_2, & r_1 < r < r_2. \end{cases}$$
 (1)

Gas mixture state and composition are assumed to be defined in arbitrary point by mixture pressure p, gas mixture temperature  $T_{\rm g}$ , mass average velocity  $\mathbf{u}_{\rm g}$  and component concentrations (molar  $C_i$  or mass  $c_i$ ) as well as by the equations of state for the mixture  $\rho_{\rm g} = pM/RT_{\rm g}$ . Average molar mass M is defined through component concentrations  $1/M = \sum_i c_i/M_i$ .

The carcass state is defined by temperature field  $T_s$ . For given carcass temperature field  $T_s$  gas parameters are described by stationary equation of continuity

$$\nabla(\rho_{\sigma}\mathbf{u}_{\mathbf{g}}) = 0 \tag{2}$$

and filtration equation for the gas mixture

$$-\nabla p = \frac{\mu}{k} \mathbf{u}_{g} + \frac{\rho_{g}}{\tilde{k}} |\mathbf{u}_{g}| \mathbf{u}_{g}. \tag{3}$$

The above two equations can be reduced to the equation for the pressure

$$\nabla \left( \frac{\nabla p}{(\mu R T_{g}/Mkp) + (|\mathbf{u}_{g}|/\widetilde{k})} \right) = 0. \tag{4}$$

Equation of mass conservation for each mixture component is

$$\nabla(\rho_{\mathbf{g}}\mathbf{u}_{\mathbf{g}}c_{i}) - \nabla(\rho_{\mathbf{g}}\mathbf{D}\otimes\nabla c_{i}) = \dot{\rho}_{i},\tag{5}$$

where  $\dot{\rho}_i$  is mass generation of *i*th component due to chemical reactions, to add  $\sum_i \dot{\rho}_i = 0$ ;  $\mathbf{D} = D_{\rm g}\mathbf{I} + \mathbf{D}_{\rm d}$ , where  $\mathbf{D}_{\rm d}$  is a dispersion diffusion tensor that depends on gas velocity  $\mathbf{u}_{\rm g}$ . It is expressed through longitudinal  $D_{\rm p}$  and transverse  $D_{\rm t}$  components as follows:

$$\mathbf{D}_{\mathrm{d}} = \begin{bmatrix} D_{\mathrm{p}}\tau_{\mathrm{z}}^{2} + D_{\mathrm{t}}\tau_{\mathrm{r}}^{2} & (D_{\mathrm{p}} - D_{\mathrm{t}})\tau_{\mathrm{z}}\tau_{\mathrm{r}} \\ (D_{\mathrm{p}} - D_{\mathrm{t}})\tau_{\mathrm{z}}\tau_{\mathrm{r}} & D_{\mathrm{p}}\tau_{\mathrm{r}}^{2} + D_{\mathrm{t}}\tau_{\mathrm{z}}^{2} \end{bmatrix}, \quad \boldsymbol{\tau} = \frac{\mathbf{u}_{\mathrm{g}}}{|\mathbf{u}_{\mathrm{g}}|}. \quad (6)$$

Energy equation for a gas

$$c_{\rm p} \nabla (\rho_{\rm g} \mathbf{u}_{\rm g} T_{\rm g}) - \nabla (\mathbf{\Lambda} \otimes \nabla T_{\rm g}) = \frac{\alpha_{\rm vol}}{m} (T_{\rm s} - T_{\rm g}) - H \dot{\rho}_{\rm 1}$$
 (7)

includes heat exchange with the carcass and energy generation due to fuel combustion. Here  $\dot{\rho}_1$  is fuel mass consumption at combustion, and  $\Lambda$  is dispersion heat conductivity tensor defined similarly to diffusivity tensor  $\mathbf{D}$ .

Non-stationary energy equation for carcass includes heat conduction of solid and radiation conductivity components and interphase heat exchange part

$$(1-m)\rho_{\rm s}c_{\rm s}\frac{\partial T_{\rm s}}{\partial t} - \nabla(\lambda\nabla T_{\rm s}) = \alpha_{\rm vol}(T_{\rm g} - T_{\rm s}), \tag{8}$$

$$\lambda = \lambda_{\rm s} + \frac{16}{3} \left( \frac{0.666m}{1-m} + 0.5 \right) d_0 \varepsilon_{\rm int} \sigma T_{\rm s}^3 \tag{9}$$

is effective heat conductivity in the carcass.

Boundary conditions. At initial moment gas temperature is assumed equal to  $T_0$ . Gas enters the chamber at  $T_0$ , so  $T_{\rm g}(z=0)=T_0$ . Gas heat exchange with walls is neglected, therefore,

$$(\mathbf{n} \cdot \nabla) T_{\sigma} = 0 \tag{10}$$

on the side surface and at the chamber's outlet crosssection.

Black body irradiation at inlet, outlet cross-section and burner sides is assumed

$$-\lambda \frac{\partial T_{s}}{\partial z} = \varepsilon_{\text{ext}} \sigma (T_{s}^{4} - T_{0}^{4}), \quad z = 0,$$

$$-\lambda \frac{\partial T_{s}}{\partial z} = \varepsilon_{\text{ext}} \sigma (T_{s}^{4} - T_{\text{ext}}^{4}), \quad z = L,$$

$$-\lambda \frac{\partial T_{s}}{\partial z} = \varepsilon_{\text{ext}} \sigma (T_{s}^{4} - T_{\text{ext}}^{4}), \quad r = r_{2}.$$
(11)

The following conditions for pressure are accepted:

Side surface non-permeability.

$$\partial p/\partial r = 0, \quad r = r_2.$$
 (12)

Normal outlet pressure.

$$p = p_0, \quad z = L. \tag{13}$$

The boundary condition for pressure at inlet cross-section (z = 0) is obtained by using pressure and flow rates  $G_1$ ,  $G_2$  compliance condition

$$\partial p/\partial z = \begin{cases} g_1 & \text{for } r < r_1, \\ g_2 & \text{for } r_1 < r < r_2, \end{cases}$$

where

$$g_1 \int_0^{r_1} \frac{r \, \mathrm{d}r}{(\mu R T_{\rm g}/M k p) + (|\mathbf{u}_{\rm g}|/\tilde{k})} = -\frac{\rho_{01} G_1}{\pi m}, \quad r < r_1, \tag{14}$$

$$g_2 \int_0^{r_1} \frac{r \, \mathrm{d}r}{(\mu R T_{\mathrm{g}} / M k p) + (|\mathbf{u}_{\mathrm{g}}| / \tilde{k})} = -\frac{\rho_{02} G_2}{\pi m}, \quad r_1 < r < r_2.$$
(15)

Wall non-permeability is applied for the gas components, Eq. (5)

$$(\mathbf{n} \cdot \nabla)c_i = \begin{cases} 0 & \text{for } r = r_2, \\ 0 & \text{for } z = L. \end{cases}$$
 (16)

Initial composition of gases is specified in the inlet crosssection, i.e.

$$c_i(z=0) = \begin{cases} c_{i1}, & r < r_1, \\ c_{i2}, & r_1 < r < r_2. \end{cases}$$
 (17)

To prevent uncontrollable diffusion and energy flow through the inlet cross-section, we assume

$$\mathbf{D}(z=0) = 0, \quad \mathbf{\Lambda}(z=0) = 0.$$
 (18)

# 3. Results and discussion

For numerical simulation the following primary parameters were accepted: volumetric heat exchange coefficient [6]:

$$\alpha_{\text{vol}} = \frac{\lambda_{\text{g}} 6(1 - m)}{d_0^2} \left[ 2 + 1.1 P r^{1/3} \left( \frac{\rho_{\text{g}} u_{\text{g}} d_0}{\mu} \right)^{0.6} \right]; \tag{19}$$

permeabilities [7]:  $k = (d_0^2m^2/150(1-m)^2)$ ,  $\tilde{k} = (d_0m/1.75(1-m))$ . Nitrogen diffusion, viscosity and heat conduction coefficients were used for gas phase. The following approximations with characteristic accuracy better than 5% in all temperature range were utilized:  $D_{\rm g} = 0.18(T/273)^{1.75}(p_{\rm o}/p)~{\rm cm}^2/{\rm s};~p_{\rm o} = 1.013\times10^5~{\rm Pa};$   $\lambda_{\rm g} = 1.4\times10^{-2} + 4.8\times10^{-5}T,~{\rm W/m/K};~\mu = 4.4\times10^{-7}T^{0.65}~{\rm Pa}$  s. The first-order brutto kinetics often used for FC modeling [8] is not applicable for non-premixed combustion simulation. For this reason we used the

second-order methane oxidation kinetics which was fitted to satisfy normal flame propagation velocity at atmospheric pressure d[CH<sub>4</sub>]/dt =  $-3.6 \times 10^{-10}$ [CH<sub>4</sub>][O<sub>2</sub>] exp(-15640/T), here concentrations [CH<sub>4</sub>] and [O<sub>2</sub>] have dimension cm<sup>-3</sup>. Other parameters: solid emissivity  $\varepsilon_{\rm ext}=0.4$ ,  $\varepsilon_{\rm int}=0.4$ ; tube length L=0.3 m; burner diameter  $d_2=0.038$  m; internal tube diameter  $d_1=0.023$  m; outlet pressure,  $p_0=1.013\times 10^5$  Pa, porosity m=0.4; gas molecular weight M=0.029 kg/mol; bedding particle diameter  $d_0=0.003$  m; methane heat content  $H=5.3\times 10^7$  J/kg; porous media heat conduction coefficient  $\lambda_{\rm s}=0.2$  J/m/K; longitudinal and transverse dispersion diffusivity

$$D_{\rm p} = 0.5 d_0 u_{\rm g}, \quad D_{\rm t} = 0.1 d_0 u_{\rm g}.$$
 (20)

In the standard case volumetric flow rates of methane and air were as follows:  $G_1 = 0.2$  and  $G_2 = 2$  m<sup>3</sup>/h, which corresponds to equivalence ratio  $\varphi = 0.95$ , concentration of methane at inlet is unity, concentration of oxygen in air is 0.21.

Ignition of the burner was simulated with the temperature initial distribution. The temperature of the 3 cm end portion of the carcass was 1200 K and the temperature of the rest part of carcass was 300 K. The physical time which was necessary to reach thermal steady state varied from 20 to 90 min, depending on flow

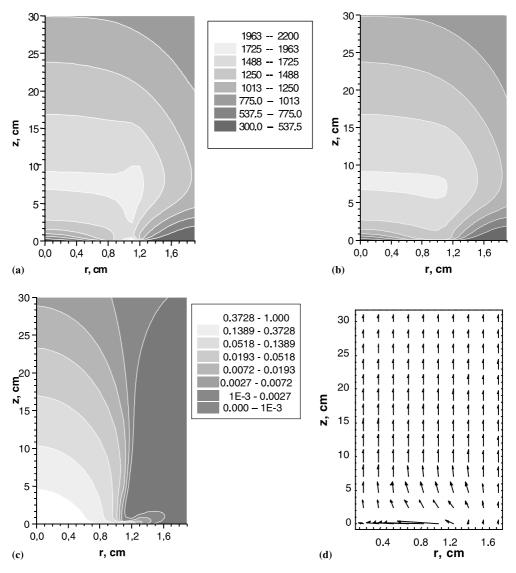


Fig. 1. Stationary temperature field of gas (a) porous media (b) methane dimentionless concentration (c) and filtration velocity field (d) as calculated for standard case parameters and free irradiation from the burner side surface.

rates and mixture equivalence ratio. Temperature, concentration and filtration fields calculated for the standard case are presented in Fig. 1. The calculations show that temperature difference between gas and solid is higher than in the case of premixed filtration combustion and may reach 200-300 K. In the area aside of the maximal heat release the gas and solid temperature fields are quite similar. On the graph in Fig. 1(c) the splash of methane concentration at the beginning of the burner is seen. This is due to the fact that dispersion provides high diffusivity for the gas at the time the temperature is not enough high to secure instantaneous methane burning out at the mixing. Fig. 1(d) demonstrates that the filtration field may be very inhomogeneous at the beginning of the burner, but quickly relax to uniform one at the length of the order of one tube diameter. To analyze heat release intensity by the length of the burner, we calculated local methane consumption rate,  $\dot{q}_1$ , Fig. 2. The logarithmic scale was applied to visualize low level effects. Calculations demonstrate that heat release zone is rather narrow and its location very slightly depends on total flow rate at the given equivalence ratio which is a proof of the diffusion character of the flame.

The radiative efficiency of the system  $\eta=Q_{\rm rad}/Q_{\rm tot}$  is defined as a ratio of irradiated energy from the surface to the total firing power. It is one of the most important parameters in the case of the radiative burner application. It was calculated as a function of the total flow rate at fixed equivalence ratio, Fig. 3, and as a function of methane flow rate at fixed air flow rate, Fig. 4. As follows from Fig. 3, the efficiency grows while total flow rate decreases. High value of the efficiency indicates very effective heat exchange and transportation from the reaction zone to the irradiating surface. Low efficiency at

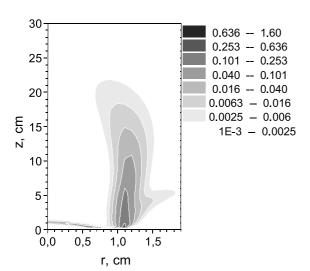


Fig. 2. Local methane consumption rate  $\dot{q}_1$ , kg/s/m<sup>3</sup>, in the burner. Logarithmic scale presentation.

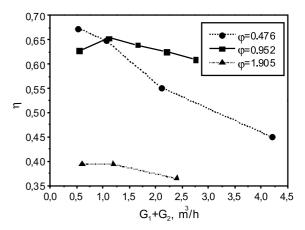


Fig. 3. Radiative efficiency of the burner as a function of total flow rate calculated for different equivalence ratios, as presented in the legend.

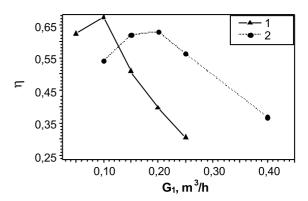


Fig. 4. Radiative efficiency of the burner as a function of methane flow rate at fixed air flow rate, as presented in the legend. Line  $1 - G_1 = 1$  m<sup>3</sup>/h, line  $2 - G_2 = 2$  m<sup>3</sup>/h.

 $\varphi=1.905$  is explained by incomplete fuel combustion. The observed (in Fig. 3) reduction of the efficiency at low flow rates is due to incomplete burning out in condition of considerable heat losses. In the case of low equivalence ratio of  $\varphi=0.476$  this effect is not exist. The efficiency peaks in Fig. 4 corresponds to the stoichiometric air–fuel mixture.

A parametric study of the radiative efficiency was performed in which the internal tube diameter and bedding particle diameter were varied. In Fig. 5 radiative efficiency presented as a function of the inner tube radius at fixed flow rates  $G_1 = 0.2 \text{ m}^3/\text{h}$ ,  $G_2 = 2.0 \text{ m}^3/\text{h}$ . The smaller inner tube corresponds to smaller radius of the heat release zone and consequently higher time of heat transfer from heat release zone to radiating surface. We expected some increase if the efficiency with the internal tube radius growth. Calculations demonstrate slight decrease of efficiency with the radius growth. This kind

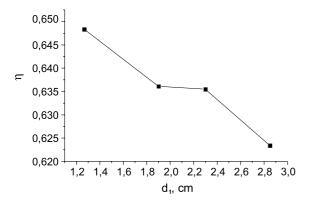


Fig. 5. Radiative efficiency dependence on the inner tube radius at fixed flow rates  $G_1 = 0.2$ ,  $G_2 = 2.0 \text{ m}^3/\text{h}$ .

of behaviour of the system is explained by incomplete fuel combustion.

The results of calculation of radiative burner efficiency as a function of the bedding particle diameter are presented in Fig. 6. One can see the maximum on the graph corresponding to the particle size  $d_0 \approx 4$  mm. This result shows that choosing optimum particle distribution may provide 2–4 percent of total efficiency of the system. Existence of the maximum may be explained in the following. Small particles corresponds to smaller hear transfer coefficient (dispersion plus radiation) and cannot provide complete transportation of the energy from reaction zone to the walls. At the same time big particle do not provide high gas-to-solid transfer coefficient, and consequently combustion products take away some more heat.

Another parameters important for radiative burner design is surface temperature distribution and heat release rate along the burner. Heat release rate may be obtained with Burke–Schumann solution [9] for the diffusion flame. As shown in [5] even the first term in the

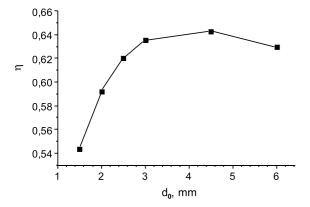


Fig. 6. Radiative efficiency dependence on the bed particle diameter  $d_0$ .  $G_1 = 0.2$ ,  $G_2 = 2.0$  m<sup>3</sup>/h.

exponential series of the Burke-Schumann solution gives good approximation for the mixing and combustion rate averaged over cross-section at coaxial feed. The corresponding fuel consumption rate  $\bar{\rho}_1 = a(1 - \exp \alpha)$  $(-14.66d_0x/r_2^2)$ ), a = normalization constant, is presented in Fig. 7 with dotted line. The same parameter was obtained numerically for two total flow rates G = 2.2and G = 1.1. Good agreement between Burke-Schumann one-exponent correlation and numerical solution for methane consumption is observed. Naturally, fuel combustion does not start from the very beginning of the burner (as Burke-Schumann model assumes) and the peak of fuel consumption lies in coordinate  $x \approx 3$  to 4 mm. Another difference is related to fuel consumption slowdown at coordinate x = 2 to 4 cm, this corresponds to local combustion intensification observable in Fig. 2. The temperature of the external boundary of the burner calculated numerically is presented on the same graph. It is seen that surface temperature distribution does not correlate to the heat release rate directly and vary with flow rate due to complex nature of heat transfer from the reaction zone to the irradiating surface. The results obtained confirmed general understanding of the process. The mixing and heat release rates are not significantly affected by flow rate variation. Surface temperature profile is shifted in the direction of filtration, maximum surface temperature and temperature variation at the length of working part of the radiative burner (~25 cm) is increased in the case of higher flow rate. The surface temperature profile shift is undesirable effect from the viewpoint of radiative burner design as far as it limits firing power range. Numerical simulation shows that this effect may be compensated with inner tube diameter increase or interchange of fuel and air feed channels.

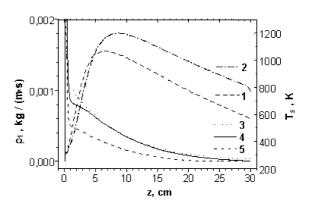


Fig. 7. Burner surface temperature (lines 1 and 2) and methane consumption rate averaged by cross-section (lines 5 and 4) calculated for total flow rates G = 1.1 and G = 2.2 m<sup>3</sup>/h, respectively. Line 3 presents one-exponent Burke–Schumann estimate for methane consumption for G = 2.2 m<sup>3</sup>/h.

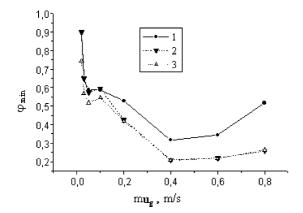


Fig. 8. Dependence of  $\varphi_{\rm min}$  on superficial gas velocity  $m{\bf u}_{\rm g}$  with different bedding particle diameter  $d_0$  and ambient temperature  $T_{\rm ext}$ . (1)  $d_0=3\times 10^{-3}$  m,  $T_{\rm ext}=300$  K, (2)  $d_0=1\times 10^{-3}$  m,  $T_{\rm ext}=300$  K, (3)  $d_0=3\times 10^{-3}$  m,  $T_{\rm ext}=500$  K.  $d_1=2\times 10^{-2}$ ,  $d_2=4\times 10^{-2}$  m is constant.

We also examined the lean limits of stable combustion in the given system. The lean limit was characterized via minimal value of total equivalence ratio providing stationary combustion in the system and was defined by numerical simulation. The length of the simulated system was L=15 cm. Superficial gas velocity  $m\mathbf{u}_{\rm g}$ , diameter of the internal and external tubes  $d_1$  and  $d_2$ , bedding particles diameter  $d_0$  and ambient temperature  $T_{\rm ext}$  were varied. Emissivity of external tube and porous carcass were accepted  $\varepsilon_{\rm ext}=0.3$ ,  $\varepsilon_{\rm int}=0.3$ , respectively. Ignition of the system was simulated by heating up of the burner to the temperature of 1200 K. The results of this investigation are presented in Figs. 8–10.

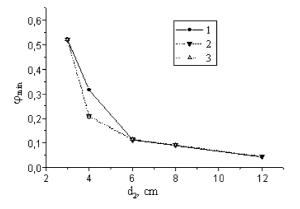


Fig. 9. Dependence of  $\varphi_{\rm min}$  on external tube diameter of  $d_2$  with different bedding particle diameter  $d_0$  and ambient temperature  $T_{\rm ext}$ . (1)  $d_0=3\times 10^{-3}$  m,  $T_{\rm ext}=300$  K, (2)  $d_0=1\times 10^{-3}$  m,  $T_{\rm ext}=300$  K, (3)  $d_0=3\times 10^{-3}$  m,  $T_{\rm ext}=500$  K.  $d_1/d_2=0.5$ ,  $m{\bf u}_{\rm g}=0.4$  m/s is constant.

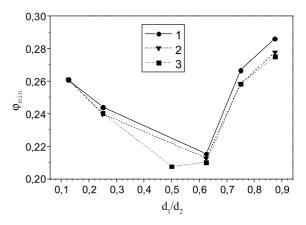


Fig. 10. Dependence of  $\varphi_{\rm min}$  on  $d_1/d_2$  with different bedding particle diameter  $d_0$ . (1)  $d_0=3\times 10^{-3}$  m,  $T_{\rm ext}=300$  K, (2)  $d_0=3\times 10^{-3}$  m,  $T_{\rm ext}=300$  K without irradiation from endwall, (3)  $d_0=1\times 10^{-3}$  m,  $T_{\rm ext}=300$  K.  $d_2=4\times 10^{-2}$  m and  $m\mathbf{u}_{\rm g}=0.4$  m/s is constant.

The dependence of  $\varphi_{\min}$  on superficial gas velocity  $m\mathbf{u}_g$  is presented in Fig. 8. Bedding particle diameter  $d_0$  and ambient temperature  $T_{\mathrm{ext}}$  were varied. Tubes diameters were  $d_1 = 2 \times 10^{-2}$ ,  $d_2 = 4 \times 10^{-2}$  m. The graph has a minimum at  $m\mathbf{u}_g \sim 0.4$  m/s. This explained by competition of the process of fuel and oxidant mixing according to Eqs. (5) and (6) and heat loses from combustion front Eqs. (7)–(9), (11) and (19). With  $m\mathbf{u}_g$  increase, intensity of mixing fuel and oxidant grows up, that, in its turn, decreases  $\varphi_{\min}$ . On the other hand,  $m\mathbf{u}_g$  increase leads to higher heat exchange rate and thus to higher heat loses from combustion front. As shown in Fig. 8, the smaller bed particles diameter  $d_0$  and lower  $T_{\mathrm{ext}}$  corresponds to lower  $\varphi_{\min}$ .

The dependence of  $\varphi_{\min}$  on the diameter of external tube  $d_2$  for the cases of different bedding particle diameters  $d_0$  and ambient temperature  $T_{\rm ext}$  is presented in Fig. 9. The tube 1 to tube 2 diameters ratio and gas velocity were as follows  $d_1/d_2=0.5$ ,  $m\mathbf{u}_{\rm g}=0.4$  m/s. The trends of the graph are explained as follows. With the  $d_2$  increase, combustion takes place farther from the surface, heat loses decrease which results in lower value of  $\varphi_{\rm min}$ . At  $d_2 \geqslant 6$ , value of  $\varphi_{\rm min}$  does not depend on  $d_0$  and  $T_{\rm ext}$  noticeably.

In Fig. 10 the  $\varphi_{\min}$  as a function of  $d_1/d_2$  is presented. The external tube diameter  $d_2=4\times 10^{-2}$  m and  $m\mathbf{u}_{\rm g}=0.4$  m/s were taken constant. The cases of different particle diameters  $d_0$  were simulated.

### 4. Conclusions

Non-premixed filtration combustion is characterized with considerable split between the gas phase and solid temperature which reaches 200–300 K in the simulated conditions and higher than the one in premixed filtration combustion systems at the same equivalence ratio and total flow rate. This effect may be utilized for implementation of high temperature gas phase chemical processes.

Simulation of NFC predicts high stability of the flame at low equivalence ratios. The proper geometry of the fuel–air feed provides the lower limit of  $\phi \approx 0.2$  for stable combustion for the simulated system.

Burke–Schumann mixing rate conforms well to heat release rate in the NFC system, although surface temperature distribution and heat transfer inside the system should be calculated with 2D model.

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